Thin-layer chromatography of 2,4-dinitrophenylosazone homologs of vicinal dicarbonyls

Recent column partition^{1,2} and adsorption³ methods have facilitated the partial resolution of 2,4-dinitrophenylosazone (DNPO) mixtures. These methods have also been supplemented by a thin-layer chromatographic (TLC) technique that will separate the osazones of vicinal dicarbonyls into classes⁴. The purpose of this communication is to describe a TLC method that effects separation of a series of DNPO



Fig. 1. Thin-layer chromatoplate of 2,4-DNPO of $C_4-C_{10} \alpha$ -ketoalkanals. Solvent front: 15 cm. Development time: 8 h.

homologs. The technique is a slight modification of a reversed-phase TLC system⁵ for the separation of homologous series of 2,4-dinitrophenylhydrazones of monocarbonyl compounds.

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NOTES

Experimental

Thin-layer plates coated with silica gel G in a 250 μ layer were dried in an oven at 100° for I h, cooled, then dipped into a solution of 10% Shell Ondina 27^{*} mineral oil in petroleum ether. Upon evaporation of the petroleum ether the plates were spotted with the DNPO mixture approximately 1/4 in. from the base of the plate in the usual manner. The plates were developed for about 7 h (or until sufficient separation had been accomplished) in a system of dioxane-water (6:4). During development, one end of the plate was exposed to the atmosphere in the manner described by LIBBEY AND DAY⁵. This technique negates multiple development.

Fig. 1 illustrates the resolution of an homologous series of α -ketoalkanals. Since the system is reversed-phase chromatography, the higher the number of carbon atoms in the parent chain the slower the migration rate. 2,3-Diketones were also resolved by the technique. In our laboratory, a mixture of the C₄, C₅, C₇ and C₈ diketone derivatives, developed for 7 h, showed resolution approximating that of the α -ketoalkanals.

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E. A. CORBIN, Anal. Chem., 34 (1962) 1244.
D. P. SCHWARTZ, J. Chromatog., 9 (1962) 187.
M. L. WOLFROM AND G. P. ARSENAULT, Anal. Chem., 32 (1960) 693.
W. Y. COBB, J. Chromatog., 14 (1964) 512.
L. M. LIBBEY AND E. A. DAY, J. Chromatog., 14 (1964) 273.

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* Note added in proof. Since submitting this note it has come to the authors' attention that Shell Oil Company no longer manufactures mineral oils. Nujol, a product of Plough, Inc., New York, has been evaluated as a substitute stationary phase and found to perform adequately. It is assumed that comparable grades of mineral oil also may be employed.

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